

This Page Is Inserted by IFW Operations  
and is not a part of the Official Record

## **BEST AVAILABLE IMAGES**

Defective images within this document are accurate representations of the original documents submitted by the applicant.

Defects in the images may include (but are not limited to):

- BLACK BORDERS
- TEXT CUT OFF AT TOP, BOTTOM OR SIDES
- FADED TEXT
- ILLEGIBLE TEXT
- SKEWED/SLANTED IMAGES
- COLORED PHOTOS
- BLACK OR VERY BLACK AND WHITE DARK PHOTOS
- GRAY SCALE DOCUMENTS

**IMAGES ARE BEST AVAILABLE COPY.**

**As rescanning documents *will not* correct images,  
please do not report the images to the  
Image Problems Mailbox.**



(12) **EUROPEAN PATENT APPLICATION**

(43) Date of publication:  
30.10.2002 Bulletin 2002/44

(51) Int Cl.<sup>7</sup>: **C11D 11/00**, C11D 3/00,  
C11D 7/36, C11D 3/37,  
C11D 7/06

(21) Application number: **02008237.6**

(22) Date of filing: **18.04.2002**

(84) Designated Contracting States:  
**AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU**  
**MC NL PT SE TR**  
Designated Extension States:  
**AL LT LV MK RO SI**

(72) Inventors:  
• **Falsina, Francesco**  
**00141 Roma (IT)**  
• **Cleuren, William G.**  
**3770 Riemst (BE)**

(30) Priority: **23.04.2001 US 840351**

(74) Representative: **HOFFMANN - EITLE**  
**Patent- und Rechtsanwälte**  
**Arabellastrasse 4**  
**81925 München (DE)**

(71) Applicant: **BetzDearborn Inc**  
**Trevose, PA 19053-6783 (US)**

(54) **Anti-scuffing treatment for alkaline bottle washing systems**

(57) The present invention is related to the addition of a specific combination of components that can significantly reduce the adverse impact of the caustic wash solution on the bottle surface of bottles being washed for reuse without compromising the cleaning activity of the wash solutions. The addition of a combination of a

polycarboxylic acid and a phosphonic acid to a caustic wash solution significantly decreases the weight loss observed in glass bottles being subjected to a caustic wash procedure.

**Description****Field of the Invention**

5 [0001] The present invention relates to improved alkaline cleaning solutions used in bottle washing systems. More particularly, the present invention relates to additives for alkaline bottle washing solutions, which are effective at reducing chemical attack and the resulting bottle material loss or scuffing which occurs in the washing of bottles with alkaline cleaning solutions.

10 **Background of the Invention**

[0002] The use of caustic solutions to wash glass bottles and other reusable food and beverage containers is widespread. For example, glass bottles used for soft drinks and beer are typically washed in an automated system with a caustic wash solution. In many areas, the use of such caustic solutions is controlled by law or industry requirements.

15 The caustic solutions used in such operations can attack the glass surface causing opacity, scratching and loss of material. This phenomenon reduces the number of times the bottles can be reused.

**Detailed Description of the Invention**

20 [0003] The washing of glass bottles for reuse is a well-known practice. For example, the washing of glass soft drink and beer bottles for reuse has been well known for many years. Such bottle washing systems are often highly automated and employ a high temperature caustic solution to wash the bottles. In addition to washing the bottles, the high temperature caustic solution can also attack the surfaces of the bottles. This results in scuffing and scratching of the surfaces, which will eventually render the bottles unusable from an aesthetic standpoint. In addition, the high temperature caustic solution can also cause loss of glass resulting in a thinner bottle. Again, this reduces the useful life of the bottles and leads to higher costs associated with un-salable dropout during distribution.

25 [0004] The caustic solutions employed in such bottle washing operations typically comprise 1% to 10% (all percentages stated herein are in weight percent) of caustic, typically, sodium hydroxide. The wash solution may also include antifoaming agents and/or metal corrosion inhibiting additives.

30 [0005] The present invention is directed to the discovery that addition of a specific combination of components can significantly reduce the adverse impact of the caustic wash solution on the glass bottle surface without compromising the cleaning activity of the wash solutions. It was discovered that the addition of a combination of a polycarboxylic acid and a phosphonic acid to a caustic wash solution significantly decreases the weight loss observed in glass beer bottles being subjected to a caustic wash procedure.

35 [0006] The anti-scuffing treatment of the present invention comprises a combination of at least one polycarboxylic acid, the sodium and/or potassium salts thereof and at least one phosphonic acid, the sodium and/or potassium salts thereof. The at least one polycarboxylic acid is preferably polyacrylic acid, having a molecular weight of from about 500-10,000. The at least one phosphonic acid is preferably a phosphonic acid derivative such as hydroxyethylidene diphosphonic acid (HEDP), amino tri(methylene phosphonic acid) (ATMP), 2-phosphono-butane-1,2,4-tricarboxylic acid (PBTC) and hydroxyphosphonoacetic acid (HPA). The most preferred phosphonic acid derivative is HEDP which, like polyacrylic acid, is a "food grade" additives. Food grade additives are acceptable additives for systems handling and treating foods or food containers.

40 [0007] The ratio of polycarboxylic acid to phosphonic acid of the present invention can range from about 1:1 to about 5:1 it is preferably about 2:1. The concentration of the polycarboxylic acid component in the caustic bottle wash solution can range from about 0.0058% to about 0.0233% (w/w). The concentration of the phosphonic acid component in the caustic wash solution can range from about 0.0031% to about 0.0124% (w/w). The treatment of the present invention can be supplied in any convenient form. The treatment components may be supplied in combination or individually. The combination of the present invention is added to a conventional aqueous caustic bottle washing solution.

45 [0008] In a preferred embodiment, the treatment is supplied as an aqueous concentrated solution comprising about 10 to 15 % (w/w) of polyacrylic acid and 5 to 8 % (w/w) of HEDP more preferably, about 12% w/w polyacrylic acid and about 6% w/w HEDP. The preferred concentrated treatment solution can be added to an aqueous caustic wash solution in concentrations ranging from about 0.05% to about 0.2% and preferably from about 0.05% to 0.1%.

50 [0009] It has been found that the addition of the antiscuffing treatment combination of the present invention can significantly reduce the weight loss of glass bottle being cleaned in a hot caustic wash system as well as reduce scuffing and opacity thereby increasing the effective life of the bottle.

55 [0010] The present invention will now be described with respect to specific examples, which are exemplary only, and not to be interpreted as limiting the scope of the attached claims.

Examples

**[0011]** Glass bottles from a beer brewery were employed in the testing. Weight loss of the bottles placed in an alkaline cleaning solution at 80°C for 24 hours was measured as a representation of the scuffing effect. The makeup water was prepared by diluting local tap water with demineralized water to achieve a calcium hardness equal to 180 ppm as CaCO<sub>3</sub>. The caustic cleaning solution was either 2% (w/w) of NaOH or 2% (w/w) NaOH and 1.5% (w/w) of any Na<sub>2</sub>CO<sub>3</sub>. Beakers of this test solution were prepared with varying concentrations of an antiscuffing treatment comprising a 2:1 ratio of polyacrylic acid, molecular weight of between 100 and 1800 and HEDP. The labels were removed from the bottles and the bottles rinsed in demineralized water, dried overnight in an oven at 105° C and cooled to ambient temperatures before weighing. A pair of bottles was immersed vertically in the test solution. The solutions were covered and maintained at 80° C while gently stirred with a magnetic stirrer for 24 hours. The bottles were removed from the test solutions, washed and rinsed with demineralized water, dried overnight at 105° C cooled to room temperature and weighed.

Example 1

**[0012]** A test solution of 2% (w/w) NaOH and 1.5% (w/w) Na<sub>2</sub>CO<sub>3</sub> was treated with 0.1% (w/w) and 0.2% (w/w) of the preferred combination treatment solution described above comprising about 12% (w/w) polyacrylic acid and about 6% (w/w) HEDP. Table 1 summarizes the weight loss results.

TABLE 1

NaOH (2% w/w) and Na <sub>2</sub> CO <sub>3</sub> (1.5% w/w)				
Bottle No.	Treatment (% w/w)	Weight before (gram)	Weight After (gram)	Weight Loss (gram)
1	--	173.9003	173.1300	0.7703
2	--	174.0011	173.2313	0.7698
3	0.1	173.8131	173.3798	0.4333
4	0.1	173.6950	173.2735	0.4215
5	0.2	174.6163	174.2115	0.4048
6	0.2	174.5110	174.1092	0.4018

**[0013]** Testing was continued with bottles 1, 2, 5, and 6. Bottles 2 and 6 were placed in a test solution as described in Table 1. Bottle 1 was placed in a test solution containing 0.2% w/w of the preferred treatment combination solution and bottle 5 placed in a solution without treatment. This was done to observe bottles 1 and 5 for any residual or carry over effect from the previous testing. Table 2 summarizes the weight loss results.

TABLE 2

NaOH (2% w/w) and Na <sub>2</sub> CO <sub>3</sub> (1.5% w/w)		
Bottle No.	Treatment (% w/w)	Weight Loss (gram)
2	--	0.7051
6	--	0.6970
1	0.2	0.0025
5	0.2	0.0037

**[0014]** The data in Table 2 shows that there was no residual or carry over effect.

**[0015]** The testing was repeated with the same test solutions as set out in Table 1 above. Table 3 summarizes the results. Again, the testing confirmed that for bottles 2 and 5, there was no residual or carry over effect.

TABLE 3

NaOH (2% w/w) and Na <sub>2</sub> CO <sub>3</sub> (1.5% w/w)		
Bottle No.	Treatment (% w/w)	Weight Loss (gram)
1	--	0.6973
2	--	0.7065
3	0.1	0.0046
4	0.1	0.0056
5	0.2	0.0741
6	0.2	0.0733

[0016] The test results summarized in Table 1, 2, and 3 show that for untreated wash solutions the weight loss is relatively constant. For wash solutions treated with the treatment combination of the present invention, while there is some variation, in all cases there was a significant decrease in the weight loss for the bottles.

#### Example 2

[0017] Testing was undertaken in a wash solution of 2% (w/w) NaOH both with and without varying concentrations of the treatment of combination of the present invention described above. The procedure described above in Example 1 was used. Table 4 summarizes the results.

TABLE 4

NaOH (2% w/w)		
Bottle No.	Treatment (% w/w)	Weight Loss (gram)
1	--	0.1693
2	--	0.1745
3	0.05	0.0012
4	0.05	0.0020
3	0.1	0.0038
4	0.1	0.0044
5	0.2	0.1921
6	0.2	0.1904
4	0.4	0.1728
6	0.4	0.1641
5	0.8	0.3002
6	0.8	0.3134

[0018] The Examples show that low dosages of the treatment combination of the present invention can significantly reduce weight loss of glass bottles being cleaned in a hot caustic wash solution.

[0019] While the present invention has been described with respect to particular embodiments thereof, it is apparent that numerous other forms and modifications of this invention will be obvious to those skilled in the art. The appended claims and this invention generally should be construed to cover all such obvious forms and modification, which are within a true spirit and scope of the present invention.

#### Claims

1. A method cleaning glass bottles comprising washing said glass bottles in an aqueous wash solution comprising:

- a. from about 0.0058% to about 0.0233% (w/w) of at least one polycarboxylic acid, the sodium and/or potassium salts thereof;
  - b. from about 0.0031% to about 0.0124% (w/w) of at least one phosphonic acid, the sodium and/or potassium salts thereof;
  - 5 c. from about 1 to about 10% (w/w) caustic; and
  - d. the balance water.
2. The method of claim 1 wherein the ratio of a:b ranges from about 1:1 to about 5:1.
  - 10 3. The method of claim 1 wherein the ratio of a:b is about 2:1.
  4. The method of claim 1 wherein said polycarboxylic acid is polyacrylic acid.
  5. The method of claim 4 wherein the molecular weight of said polyacrylic acid ranges from about 500 to about 10,000.
  - 15 6. The method of claim 1 wherein said phosphonic acid is a phosphonic acid derivative selected from the group consisting of hydroxyethylidene diphosphonic acid, amino tri(methylene phosphonic acid), 2-phosphono-butane-1,2,4-tricarboxylic acid and hydroxyphosphonoacetic acid.
  - 20 7. The method of claim 1 wherein said phosphonic acid is hydroxyethylidene diphosphonic acid.
  8. A method of inhibiting the scuffing of glass bottles being washed with an aqueous caustic wash solution comprising adding to said aqueous caustic was solution from about 0.0058% to about 0.0233% (w/w) of at least one polycarboxylic acid, the sodium and/or potassium salts thereof and from about 0.0031% to about 0.0124 % (w/w) of at least one phosphonic acid.
  - 25 9. The method of claim 8 wherein the ratio of polycarboxylic acid to phosphonic acid ranges from about 1:1 to about 5:1.
  - 30 10. The method of claim 8 wherein the ratio of polycarboxylic acid to phosphonic acid is about 2:1.
  11. The method of claim 8 wherein said polycarboxylic acid is polyacrylic acid.
  12. The method of claim 11 wherein the molecular weight of said polyacrylic acid ranges from about 500 to about 10,000.
  - 35 13. The method of claim 8 wherein said phosphonic acid is a phosphonic acid derivative selected from the group consisting of hydroxyethylidene diphosphonic acid, amino tri(methylene phosphonic acid), 2-phosphono-butane-1,2,4-tricarboxylic acid and hydroxyphosphonoacetic acid.
  - 40 14. The method of claim 8 wherein said phosphonic acid is hydroxyethylidene diphosphonic acid.
  15. An aqueous concentrated solution comprising about 10 to 15 % (w/w) of at least one polycarboxylic acid and 5 to 8 % (w/w) of at least one phosphonic acid.
  - 45 16. The use of the aqueous concentrated solution as claimed in claim 15 in the preparation of an aqueous wash solution for cleaning or inhibiting the scuffing of glass bottles.

50

55



European Patent  
Office

# EUROPEAN SEARCH REPORT

Application Number  
EP 02 00 8237

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
X	WO 01 12769 A (UNILEVER PLC ; LEVER HINDUSTAN LTD (IN); UNILEVER NV (NL)) 22 February 2001 (2001-02-22) * examples 5,6 * * page 9, line 5 - line 11 * * page 9, line 22 - line 23 * * page 10, line 20 - line 24 * * page 11, line 5 - line 6 *	1-7	C11D11/00 C11D3/00 C11D7/36 C11D3/37 C11D7/06
X	GB 2 203 163 A (SANDOZ LTD) 12 October 1988 (1988-10-12) * examples 4-6; tables 1,2 * * page 4, line 20 - line 25 *	1-7	
X	US 6 106 633 A (ROUILLARD CAROL ANNE) 22 August 2000 (2000-08-22)	1,2,4-6	
Y	* column 7, line 55 - line 60; claim 1 * * tables 4, FORMULA, 8 *	8-14	
Y	WO 96 16159 A (UNILEVER NV ; UNILEVER PLC (GB)) 30 May 1996 (1996-05-30) * page 3, line 1 - line 11 * * claims; example 1 *	8-14	TECHNICAL FIELDS SEARCHED (Int.Cl.7) C11D
X	GB 2 227 021 A (SANDOZ LTD) 18 July 1990 (1990-07-18) * tables 2, EX.11-12 *	15,16	
X	US 5 849 095 A (ROUILLARD CAROL) 15 December 1998 (1998-12-15) * column 3, line 22 - line 23 * * column 3, line 35 - line 55 * * column 4, line 35 - line 49; claim 1; table 7 *	1-7	
Y		8-14	
The present search report has been drawn up for all claims			
Place of search: MUNICH		Date of completion of the search: 27 June 2002	Examiner: Loloiu, C
<p>CATEGORY OF CITED DOCUMENTS</p> <p>X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document</p> <p>T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons &amp; : member of the same patent family, corresponding document</p>			

EPO FORM 1503 (03.02) (P4/C01)

**ANNEX TO THE EUROPEAN SEARCH REPORT  
ON EUROPEAN PATENT APPLICATION NO.**

EP 02 00 8237

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on  
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

27-06-2002

Patent document cited in search report		Publication date		Patent family member(s)	Publication date
WO 0112769	A	22-02-2001	US	6367487 B1	09-04-2002
			AU	6268800 A	13-03-2001
			BR	0013102 A	30-04-2002
			WO	0112769 A1	22-02-2001
			EP	1203067 A1	08-05-2002
GB 2203163	A	12-10-1988	CH	675881 A5	15-11-1990
			DE	3810107 A1	13-10-1988
			FR	2613378 A1	07-10-1988
			IT	1219894 B	24-05-1990
			JP	63273700 A	10-11-1988
US 6106633	A	22-08-2000	US	5849095 A	15-12-1998
			AU	728702 B2	18-01-2001
			AU	2029397 A	29-10-1997
			BR	9708539 A	03-08-1999
			CA	2251708 A1	16-10-1997
			WO	9738079 A1	16-10-1997
			EP	0892848 A1	27-01-1999
			JP	11514021 T	30-11-1999
			ZA	9702976 A	08-10-1998
WO 9616159	A	30-05-1996	AU	4173596 A	17-06-1996
			BR	9509693 A	14-10-1997
			DE	69516537 D1	31-05-2000
			DE	69516537 T2	10-08-2000
			WO	9616159 A1	30-05-1996
			EP	0793711 A1	10-09-1997
			ES	2145937 T3	16-07-2000
			FI	972124 A	19-05-1997
			ZA	9509795 A	19-05-1997
GB 2227021	A	18-07-1990	BE	1003581 A5	28-04-1992
			CH	679311 A5	31-01-1992
			DE	3943139 A1	12-07-1990
			FR	2641544 A1	13-07-1990
			IT	1239740 B	15-11-1993
			JP	2228399 A	11-09-1990
			NL	9000040 A	01-08-1990
US 5849095	A	15-12-1998	US	6106633 A	22-08-2000
			AU	728702 B2	18-01-2001
			AU	2029397 A	29-10-1997
			BR	9708539 A	03-08-1999
			CA	2251708 A1	16-10-1997
			WO	9738079 A1	16-10-1997

EPO FORM P0159

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82



